Tetrahedron Letters 49 (2008) 4513-4515

Contents lists available at ScienceDirect

Tetrahedron Letters

journal homepage: www.elsevier.com/locate/tetlet



A diene-transmissive Diels–Alder reaction involving inverse electron-demand hetero-Diels–Alder cycloaddition of cross-conjugated azatrienes

Satoru Kobayashi, Tomoki Furuya, Takashi Otani, Takao Saito*

Department of Chemistry, Faculty of Science, Tokyo University of Science, Kagurazaka, Shinjuku-ku, Tokyo 162-8601, Japan

ARTICLE INFO

Article history: Received 10 April 2008 Revised 6 May 2008 Accepted 12 May 2008 Available online 15 May 2008

ABSTRACT

The initial inverse electron-demand hetero-Diels–Alder reaction of *N*-sulfonyldivinylmethanimine with electron-rich dienophiles (ethyl vinyl ether and ethyl vinyl sulfide) affords [4+2] cycloadducts with high *endo* selectivity. The monocycloadducts then undergo a second Diels–Alder reaction on the newly formed diene unit with electron-deficient dienophiles (tetracyanoethylene, 4-phenyl-1,2,4-triazoline-3,5-dione, and *N*-phenylmaleimide) to give highly stereoselectively the crossed biscycloadducts, hexa- and octa-hydroquinolines, and octahydropyridopyridazines.

© 2008 Elsevier Ltd. All rights reserved.

The diene-transmissive Diels–Alder (DTDA) reaction is a useful and attractive method for constructing polyring-fused cyclic compounds, consisting of two sequential (tandem) Diels–Alder (DA) reactions that involve an initial DA reaction of a cross-conjugated triene (or its equivalent), followed by a second DA reaction of the monoadduct using the newly formed diene unit. Many advances and applications of this DTDA methodology of carbotrienes have been reported.¹ However, only a few examples of the dienetransmissive hetero-Diels–Alder (DTHDA) reaction involving one or more hetero-DA reactions in the tandem sequence have been reported, despite its high potential for straightforward and efficient construction of polyring-fused heterocyclic compounds with high regio- and stereoselectivity.^{2–5}

The first reported examples of the DTHDA reaction included that of cross-conjugated thiatrienes.^{2,3} Tsuge et al.⁴ and Spino et al.⁵ demonstrated the DTHDA reaction of cross-conjugated oxatrienes. Our group previously reported the first DTHDA reaction of cross-conjugated azatrienes to produce hexa- and octahydroquinazolinones with high regio- and stereoselectivities, which included the initial aza-Diels-Alder reaction with tosyl isocyanate.⁶ We also succeeded in a stereoselective synthesis of hexahydroquinolinones by a DTHDA methodology using ketenes as a dienophile in the initial aza-DA reaction.⁷ To extend this azatriene-DTHDA methodology, we were prompted to use cross-conjugated azatrienes bearing an electron-withdrawing sulfonyl group on the nitrogen atom,⁸ which involved an inverse electron-demand aza-DA reaction in the initial cycloaddition. We report the preliminary results here.

The cross-conjugated azatrienes 1 were prepared by condensation between di-β-styryl ketone and corresponding sulfonamides using TiCl₄ and Et₃N; and they were stable enough to allow isolation after aqueous workup and/or column chromatography. First, the initial DA reaction of the azatrienes **1** with ethyl vinyl ether was performed (Scheme 1, Table 1). When N-methanesulfonyl azatriene 1a was heated in the presence of excess ethyl vinyl ether in toluene for 23 h, the corresponding [4+2] cycloadduct **2a**⁹ was obtained in 68% yield with an endo:exo ratio of 90:10 (entry 1). Azatrienes 1b and 1c also reacted with ethyl vinyl ether under the same reaction conditions to give highly endo selectively 2b and 2c in 86% and 88% yields, respectively; no exo-isomers were detected in the crude mixture (entries 2 and 3). Similarly, the reactions of 1a and 1b with ethyl vinyl sulfide proceeded in refluxing toluene to produce the monoadducts 3a and 3b in fair to good yields with high endo selectivity (entries 4 and 5).

Because the obtained monoadducts 2 and 3 possess an electronrich aminodiene moiety, a second DA reaction could be performed with electron-deficient dienophiles. First, the reactions of 2 and 3 with tetracyanoethylene (TCNE) were carried out to examine diastereo- π -facial selectivity (Scheme 2 and Table 2). The reaction of 2a-c with TCNE proceeded at room temperature for 10 min to produce [4+2] cycloadducts $4a-c^{10}$ in 93–99% yields with complete diastereo- π -facial selectivity (Table 2, entries 1–3). Similarly, the reaction of **3a**,**b** proceeded smoothly in refluxing dichloromethane to give **5a**,**b** in high yields (entries 4 and 5). A one-pot procedure $1b \rightarrow 2b \rightarrow 4b$ proved the DTHDA methodology to be even more effective and viable (entry 6). In all cases, the dienophile added from the less-hindered bottom H-4-side of the diene moiety, a conclusion that was supported by the fact that the large vicinal coupling constant (ca. 12 Hz) between H-4 and H-4a in the bisadducts (4 and 5) indicated a trans diaxial relationship. In the

^{*} Corresponding author. Tel.: +81 3 5228 8254; fax: +81 3 5261 4631. *E-mail address:* tsaito@rs.kagu.tus.ac.jp (T. Saito).

^{0040-4039/\$ -} see front matter @ 2008 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetlet.2008.05.054



Scheme 1.

Table 1Initial cycloaddition of cross-conjugated azatrienes 1

Entry	Azatriene	R	Х	Time (h)	Adduct	Yield (%)	endo:exo ^a
1	1a	Me	OEt	23	2a	68	90:10
2	1b	p-Tol	OEt	15	2b	86	>95:5
3	1c	Ph	OEt	13	2c	88	>95:5
4	1a	Me	SEt	48	3a	68	>95:5
5	1b	p-Tol	SEt	37	3b	51	>95:5







Table 2Second cycloaddition with TCNE

Entry	R	Х	Diene	Conditions	Adduct	Yield (%)
1	Me	EtO	2a	rt, 10 min	4a	93
2	p-Tol	EtO	2b	rt, 10 min	4b	95
3	Ph	EtO	2c	rt, 10 min	4c	99
4	Me	EtS	3a	40 °C, 30 min	5a	76
5	p-Tol	EtS	3b	40 °C, 60 min	5b	93
6	p-Tol	EtO	2b	110 °C, 15 h→rt, 2 d	4b	76

endo-monoadduct, the vicinal coupling constants between H-3' and H-4 and between H-3 and H-4 were observed to be ca. 8 Hz and 5–6 Hz, respectively, suggesting that H-4 orients in a quasi-equatorial position, and hence the phenyl substituent takes a quasi-axial position. Therefore, the top side of the tetrahydropyridine ring was blocked by the more bulky phenyl group from attack by the dienophile.

The reactions of **2** and **3** with 4-phenyl-1,2,4-triazoline-3,5dione (PTAD) were also examined to confirm diastereo- π -facial selectivity (Scheme 3, Table 3). Dienes **2a–c** and **3a,b** reacted rapidly within 10 min at room temperature to produce the



Scheme 3.

 Table 3

 Second cycloaddition with PTAD

Entry	R	Х	Diene	Adduct	Yield (%)
1	Me	EtO	2a	6a	99
2	p-Tol	EtO	2b	6b	99
3	Ph	EtO	2c	6c	99
4	Me	EtS	3a	7a	99
5	p-Tol	EtS	3b	7b	92

cycloadducts **6a–c** and **7a,b**, respectively, in quantitative yields. The reaction was highly diastereo– π -face selective, the same as the reaction with TCNE.

The second DA reaction with *N*-phenylmaleimide (*N*-PhMI) was carried out to examine *endo/exo* selectivity in addition to diaste-reo- π -facial selectivity (Scheme 4, Table 4). The monoadducts **2a**–**c** reacted with *N*-PhMI in refluxing toluene to produce cycload-ducts **9a** or **8b,c** as final products. The reaction proceeded with high *endo* and π -facial selectivities to give initially the 1:1-cycload-duct, from which elimination of ethanol occurred to furnish compounds **8b,c** or, in the case of entry 1, **9a** after H-migration from **8a**.

In conclusion, the diene-transmissive hetero-Diels–Alder reaction of N-sulfonylated cross-conjugated azatrienes including an inverse electron-demand aza-Diels–Alder reaction in the initial



 Table 4

 Second cycloaddition with N-PhMI

Entry	R	Diene	Time (h)	Adduct	Yield (%)
1 2	Me p-Tol	2a 2b 2a	3 18	9a 8b	34 22

cycloaddition step has been developed. The protocol provides a new entry to the highly stereoselective synthesis of octahydroquinolines and pyridopyridazines. Further work to extend the scope of this methodology is currently under way.

References and notes

- 1. For DT(H)DA of carbotrienes: (a) Blomquist, A. T.; Verdol, J. A. J. Am. Chem. Soc. 1955, 77, 81; (b) Bailey, W. J.; Economy, J. J. Am. Chem. Soc. 1955, 77, 1133; (c) Tsuge, O.; Wada, E.; Kanemasa, S. Chem. Lett. 1983, 239; (d) Tsuge, O.; Kanemasa, S.; Wada, E.; Sakoh, H. Yuki Gosei Kagaku Kyokaishi 1986, 44, 756. and references cited therein; (e) Tsuge, O.; Hatta, T.; Yakata, K.; Maeda, H. Chem. Lett. 1994, 1833; (f) Hosomi, A.; Masunari, T.; Tominaga, Y.; Yanagi, T.; Hojo, M. Tetrahedron Lett. 1990, 31, 6201; (g) Adam, W.; Deufel, T.; Finzel, R.; Griesbeck, A. G.; Hirt, J. J. Org. Chem. 1992, 57, 3991; (h) Woo, S.; Squire, N.; Fallis, A. G. Org. Lett. 1999, 1, 573; (i) Woo, S.; Legoupy, S.; Parra, S.; Fallis, A. G. Org. Lett. 1999, 1, 1013; (j) Kwon, O.; Park, S. B.; Schreiber, S. L. J. Am. Chem. Soc. 2002, 124, 13402; (k) Payne, A. D.; Willis, A. C.; Sherburn, M. S. J. Am. Chem. Soc. 2005, 127, 12188; (1) Brummond, K. M.; You, L. Tetrahedron 2005, 61, 6180; (m) Mitasev, B.; Yan, B.; Brummond, K. M. Heterocycles 2006, 70, 367; (n) Bradford, T. A.; Payne, A. D.; Willis, A. C.; Paddon-Row, M. N.; Sherburn, M. S. Org. Lett. 2007, 9, 4861; (o) Miller, N. A.; Willis, A. C.; Paddon-Row, M. N.; Sherburn, M. S. Angew. Chem., Int. Ed. 2007, 46, 937; (p) Souweha, M. S.; Arab, A.; ApSimon, M.; Fallis, A. G. Org. Lett. 2007, 9, 615; (q) Souweha, M. S.; Enright, G. D.; Fallis, A. G. Org. Lett. 2007, 9, 5163.
- 2. Motoki, S.; Matsuo, Y.; Terauchi, Y. Bull. Chem. Soc. Jpn. 1990, 63, 284.
- Saito, T.; Kimura, H.; Sakamaki, K.; Karakasa, T.; Moriyama, S. Chem. Commun. 1996, 811.
- (a) Tsuge, O.; Hatta, T.; Yoshitomi, H.; Kurosaka, K.; Fujiwara, T.; Maeda, H.; Kakehi, A. *Heterocycles* **1995**, *41*, 225; (b) Tsuge, O.; Hatta, T.; Fujiwara, T.; Yokohari, T.; Tsuge, A. *Heterocycles* **1999**, *50*, 661.
- 5. (a) Spino, C.; Liu, G. J. Org. Chem. 1993, 58, 817; (b) Spino, C.; Liu, G.; Tu, N.; Girard, S. J. Org. Chem. 1994, 59, 5596; (c) Spino, C.; Hill, B.; Dubé, P.; Gingras, S. Can. J. Chem. 2003, 81, 81; (d) Dion, A.; Dubé, P.; Spino, C. Org. Lett. 2005, 7,

5601; (e) Spino, C. Synlett **2006**, 23; (f) Perreault, S.; Spino, C. Org. Lett. **2006**, 8, 4385.

- Saito, T.; Kimura, H.; Chonan, T.; Soda, T.; Karakasa, T. Chem. Commun. 1997, 1013.
- Saito, T.; Kobayashi, S.; Ohgaki, M.; Wada, M.; Nagahiro, C. Tetrahedron Lett. 2002, 43, 2627.
- Boger et al. showed that an N-sulfonyl group on an azadiene was a quite effective substituent for an inverse electron-demand aza-DA reaction: (a) Boger, D. L.; Curran, T. T. J. Org. Chem. **1990**, 55, 5439; (b) Boger, D. L.; Corbett, W. L.; Curran, T. T.; Kasper, A. M. J. Am. Chem. Soc. **1991**, *113*, 1713; (c) Boger, D. L. J. Am. Chem. Soc. **2006**, *128*, 2587 and references cited therein.
- Compound **2a** (*endo*, H^2-H^4 *cis*): Colorless crystals; mp 111–113 °C; IR (KBr): 2970, 1335, 1157, 1119 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.14 (t, *J* = 7.1 Hz, 9 3H, OCH₂CH₃), 2.22 (ddd, J = 3.5, 4.5, 14.2 Hz, 1H, H-3), 2.51 (ddd, J = 4.8, 8.3, 14.2 Hz, 1 H, H-3'), 3.02 (s, 3H, Ms), 3.50 (dq, J = 7.1, 9.3 Hz, 1H, OCH₂CH₃), 3.54 (ddd, J = 3.7, 4.5, 8.3 Hz, 1H, H-4), 3.78 (dq, J = 7.1, 9.3 Hz, 1H, OCH₂CH₃), 5.54 (dd, J = 3.5, 4.8 Hz, 1H, H-2), 6.02 (d, J = 3.7 Hz, 1H, H-5), 6.88 (s, 2H, H-7, H-8), 7.18–7.36 (m, 8H, Ar), 7.42–7.45 (m, 2H, Ar); 13 C NMR (126 MHz, CDCl₃) δ 14.8 (CH₃), 30.9 (CH₃), 37.2 (CH), 37.6 (CH₂), 40.6 (CH₂), 63.4 (CH), 83.9 (CH), 122.0 (CH), 126.5 (2CH), 126.7 (2CH), 127.9 (CH), 128.2 (2CH), 128.3 (2CH), 128.6 (CH), 129.9 (CH), 134.8 (C), 136.6 (C), 144.5 (C). HRMS-ESI m/z: [M+Na]⁺ calcd for C22H25NNaO3S: 406.1447, found: 406.1461. Anal. Calcd for C22H25NO3S: C, 68.90; H, 6.57; N, 3.65. Found: C, 69.11; H, 6.79; N, 3.67. Compound 2a (exo, H²-H⁴ trans): Yellow oil; IR (neat): 1335, 1157, 1119 cm⁻¹; ¹H NMR (600 MHz, CDCl₃) δ 1.28 (t, J = 7.1 Hz, 3H, OCH₂CH₃), 1.77 (ddd, J = 2.4, 12.6, 13.9 Hz, 1H, H-3), 2.37 (dddd, J = 1.5, 2.4, 6.6, 13.9 Hz, 1H, H-3'), 3.01 (s, 3H, Ms), 3.68 (dq, J = 7.1, 9.6 Hz, 1H, OCH₂CH₃), 3.86 (dq, J = 7.1, 9.6 Hz, 1H, OCH₂CH₃), 3.88 (m, J = 1.5, 6.6, 12.6 Hz, 1H, H-4), 5.54 (dd, J = 2.4, 2.4 Hz, 1H, H-2), 5.67 (dd, J = 1.5, 7.15 Hz, 1H, H-5), 6.81 (d, *J* = 15.6 Hz, 1H, H-8), 7.01 (d, *J* = 15.6 Hz, 1H, H-7), 7.23–7.44 (m, 10H, Ar); ¹³C NMR (151 MHz, CDCl₃) δ 15.0 (CH₃), 36.1 (CH₂), 36.3 (CH), 40.5 (CH), 63.5 (CH₂), 84.3 (CH₃), 116.6 (CH), 126.6 (CH), 126.8 (2CH), 126.9 (CH), 127.5 (2CH), 127.9 (CH), 128.6 (2CH), 128.9 (2CH), 129.9 (CH), 134.1 (C), 136.6 (C), 143.7 (C). HRMS-ESI m/z: [M+Na]⁺ calcd for C₂₂H₂₅NNaO₃S: 406.1447, found: 406.1465.
- 10. Compound **4a**: Colorless crystals; mp 176–177 °C; IR (KBr): 1350, 1165 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 1.23 (t, *J* = 7.1 Hz, 3H, OCH₂CH₃), 2.31 (ddd, *J* = 2.9, 2.9, 15.1 Hz, 1H, H-3), 2.56 (ddd, *J* = 2.9, 9.7, 15.1 Hz, 1H, H-3'), 3.19 (s, 3H, Ms), 3.38 (dq, *J* = 7.1, 8.9 Hz, 1H, OCH₂CH₃), 3.53 (dq, *J* = 7.1, 8.9 Hz, 1H, OCH₂CH₃), 3.9 (ddd, *J* = 2.9, 9.7, 11.5 Hz, 1H, H-4), 4.19 (ddd, *J* = 1.8, 18, 11.5 Hz, 1H, H-44), 4.5 (dd, *J* = 1.8, 2.8 Hz, 1H, H-7), 5.44 (dd, *J* = 2.9, 2.9 Hz, 1H, H-2), 6.50 (dd, *J* = 1.8, 2.8 Hz, 1H, H-8), 7.26–7.53 (m, 10H, Ar); ¹³C NMR (126 MHz, CDCl₃) δ 14.9 (CH₃), 35.1 (CH₂), 40.2 (CH), 40.9 (CH₃), 41.3 (C), 43.2 (CH), 43.8 (C), 46.8 (CH), 64.7 (CH₂), 85.3 (CH), 108.3 (C), 109.7 (C), 111.1 (C), 111.3 (C), 114.2 (CH), 128.4 (CH), 128.6 (CH), 129.2 (2CH), 129.3 (3CH), 130.0 (2CH), 130.4 (CH), 132.1 (C), 132.5 (C), 139.7 (C). HRMS-ESI *m/z*: [M+Na]* calcd for C₂₈H₂₅N₅NaO₃S: 534.1570, found: 534.1546.